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Experimental Investigation of Performance, Combustion And Emissions of CI Engine Using Di-Ethyl Ether Additive with Honge Bio Diesel

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Abstract-Demand for fuel and the use of petroleum products are increasing day by day which causes serious problems such as petroleum depletion, environment degradation etc. So biodiesel is a good alternative for conventional diesel fuel. By using biodiesel there are also some disadvantages such as high oxides of nitrogen, high fuel consumption and higher density. Additives play an important role in enhancing the properties of biodiesel .In this context, oxygenated additives like Diethyl ether (DEE) and Ethanol are rich in oxygen content. DEE possess different properties altogether with higher cetane number and lesser autoignition temperature etc, may show the way for smoother start of combustion at lower atmospheric temperature also. In this paper DEE and Ethanol are blended with the B20 Honge and tested at different loads on DI diesel engine. Emission levels are decreased substantially with DEE and Ethanol blended with B20 Honge as compared with diesel at higher loads. The thermal efficiency will increase and specific fuel consumption is better in case of oxygenated additive blend. Both additives improve combustion characteristics and also reduce the emissions of unburnt hydrocarbons (UBHC).

Keywords—CI Engine, emission, B20 Honge, Diethyl ether, Ethanol.

1. INTRODUCTION

India is now the world's third largest importer of crude oil after the United States and China. India's import dependence is rising fast and it will be one of the largest sources of demand growth for the foreseeable future. Petroleum remains the largest source of energy, but its share of world marketed energy declines from 33% in 2015 to 31% in 2040. On a worldwide basis, liquids consumption increases in the industrial and transportation sectors, and declines in the electric power sector [1].

Environmental pollution is caused bv conventional fossil fuels while their demand is ever increasing. So this requires more attention on alternate fuels such as biodiesel and ethanolbiodiesel blends. Both biodiesel and ethanol can be synthesized from feedstock which is a renewable resource. Efforts have been made to replace petroleum based fuels with as much biofuel as possible because conventional diesel fuel by itself cannot be entirely replaced by biofuel. Transesterification is one of the processes for production of biodiesel using vegetable/animal oil with a short- chain alcohol like methanol or ethanol. This reaction leads to the formation of methyl esters which requires raw oil, 15% of methanol and 5% of sodium hydroxide on mass basis. In most cases, to attain the equilibrium

condition of the reaction the temperature is maintained below the normal boiling point methanol(65° C). Methanol as the potential to reduce both NO_x and particulate matter (PM) due to its high oxygen content and high latent heat of evaporation. Using methanol containing additives, there was a slightly decrease in PM and NO_x but where as general increase in both UHC and CO emissions. However, some of the disadvantages using methanol-diesel blends have been observed: it has reduction in viscosity, lubricity, cetane number and ignitability [2-3].

Ethanol can be easily converted into diethyl ether (DEE) by dehydration process. DEE is an excellent compression-ignition fuel with higher energy density than ethanol. It is widely known as a coldstart aid for engines. Alternatively DEE has a higher cetane number compared to the diesel fuel and higher calorific value compared to ethanol [5].

A. Transesterification

Transesterification is the process of exchanging the alkoxy group of an ester compound by a primary alcohol forming fatty acid alkyl esters (biodiesel) and glycerin. Chemically transesterification means taking a triglyceride molecule or a complex fatty acids, neutralizing the free fatty acids, removing the glycerin and creating an alcohol ester

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[4] .The main objective of this process is to reduce the viscosity of the biodiesel, as it cannot be directly mixed with the Diesel as they do not form homogenous mixture on blending.

In this mechanism, as shown in fig-1, the carbonyl carbon of the starting ester (RCCOR₁) undergoes nucleophilic attack by the incoming alkoxide (R₂O-) to give away a tetrahedral intermediate ,which reverts to starting material or proceeds to the transesterified product (RCOOR₂) [6].

$$R^{+}_{R^{+}_{-}O^{-}}R^{1} \xrightarrow{R^{+}_{-}O^{-}}R^{1} \xrightarrow{R^{+}_{-}O^{-}}R^{1} \xrightarrow{R^{+}_{-}O^{-}}R^{2} + R^{+}_{-}O^{-}$$

Fig-1: Transesterification

This can be achieved by mixing methanol with NaOH(sodium hydroxide 1% by weight) to make sodium methoxide. this mixture is gradually added to the honge oil when the temperature of honge oil reaches 75 degrees, the mixture was stirred continuously for few minutes and then 1% by volume of H_2SO_4 (Sulphuric acid) .The mixture was allowed to settle down for 24 h, so that all the the glycerol settles down and biodiesel floats above it.

BPBrake PowerBTDCBefore Top Dead CentreBTEBrake Thermal EfficiencyHRRHeat release rateB2020% biodiesel & 80% dieselCICompression IgnitionCOCarbon monoxideHOMEHonge Oil Methyl EsterEGTExhaust gas temperatureICInternal CombustionNOxOxides of Nitrogen		Nomenclature
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CICompression IgnitionCOCarbon monoxideHOMEHonge Oil Methyl EsterEGTExhaust gas temperatureICInternal CombustionNOxOxides of Nitrogen	B20	20% biodiesel & 80% diesel
COCarbon monoxideHOMEHonge Oil Methyl EsterEGTExhaust gas temperatureICInternal CombustionNOxOxides of Nitrogen	CI	Compression Ignition
HOMEHonge Oil Methyl EsterEGTExhaust gas temperatureICInternal CombustionNOxOxides of Nitrogen	CO	Carbon monoxide
EGT Exhaust gas temperature IC Internal Combustion	HOME	Honge Oil Methyl Ester
IC Internal Combustion	EGT	Exhaust gas temperature
NOv Ovides of Nitrogen	IC	Internal Combustion
Oxides of Millogen	NOx	Oxides of Nitrogen
ppm parts per million	ppm	parts per million
UBHC Unburnt hydrocarbons	UBHC	Unburnt hydrocarbons

2. EXPERIMENTAL SETUP FOR ENGINE TESTING

The performance tests were carried on a computerized Diesel engine . Single cylinder ,4 stroke , naturally aspired direct injection , water cooled diesel engine of 7.5 KW rated power at 1500 rpm was directly coupled to dynamometer. These signals are interfaced to control panel which is connected to a digital computer through engine indicator for diagrams.

The computerized test engine was used to record the various parameters such as air flow rate, temperature, fuel flow rate, etc .and for calculating the different performance characteristics such as brake thermal efficiency, brake specific fuel consumption, volumetric efficiency etc. Determination of viscosity, calorific value, density, flash point and fire point are carried out using redwood viscometer, Bomb calorimeter, Pensky apparatus. These values are fed to computerized engine software to calculate various performance factors. Similarly combustion characteristics such as heat release rate, peak pressure were also calculated.

The exhaust gas analyzer is used to find Oxides of nitrogen, Carbon monoxide and Unburnt hydrocarbon emissions in the exhaust. These instruments were used to compute the black smoke emission using extinction of a light beam by scattering and absorption. Opacity is also a measuring extinction of light between light sources and receiver and this is calculated in terms of percentage. The specifications of engine are tabulated in Table 1.

The whole set of experiments were conducted at a injection timing of 26 ⁰ BTDC and Compression ratio of 17.5:1. The tests were conducted at five distinct part load conditions namely, zero load, One Fourth Load, Half Full Load, Three Fourth Full Load and Full Load with diesel, diesel blended

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with 20% of honge oil, diesel blended with 20% of honge oil, DEE and ethanol. This was done in order to get favorable results without over stressing the engine. The engine is initially made to run at 1500rpm continuously for one hour in order to achieve the thermal equilibrium under operating conditions.

Make	Kirloskar	
Speed	1500 RPM	
Compression Ratio	17.5:1	
Cylinder Bore	87.5 mm	
Stroke	110 mm	
Connecting Rod Length	234 mm	
Cooling water Orifice diameter	20 mm	
Rated Power	7.5 KW	
Maximum Load	12 KN	
Injection Opening Pressure	180 bar	
Number of Cylinders	1	

Table I. Test Engine Specifications.

Table II. Properties Of Fuel And Blends.	Table Ii.	Properties	Of Fuel	And Blends.
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Properties	Diesel	Pongamia oil (Honge)	HOME	DEE
Density (kg/m ³) at 40 ⁰ C	828	915	873	713
Kinematic Viscosity (centistokes) at 40°C	3.78	42.78	5.546	0.23
Specific gravity (at 40 ⁰ C)	0.828	0.909	0.873	0.713
Calorific value (KJ/Kg)	44030	37304	40210	33900

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3. RESULTS AND DISCUSSION

A) Performance characteristics.

1) Brake Thermal Efficiency:

Figure 2 depicts the comparison of brake thermal efficiency of Diesel and Honge Biodiesel blends at various loads. The graph exhibits that the brake thermal efficiency of Honge biodiesel blend (B20 Honge + DEE + Ethanol) is higher than that of Diesel at various loads. It can be observed from the graph that at low loads all the fuel blends deliver same thermal efficiencies[7]. The validation for the Honge biodiesel blend (B20 Honge + DEE+

Ethanol) showing higher brake thermal efficiency is due to the high oxygen content in the blends of Honge as ethanol and DEE consists of high oxygen content which readily helps in combustion of hydrocarbons[11]. During specific injection timing as the load increases there is a linear increment in the Brake thermal efficiency and this happens due to increase in fuel conversion efficiency and better Oxygen – fuel ratios at high load. Thus this attributes to the increase in the combustion efficiency of the biodiesel. The average Brake thermal efficiency obtained is 36%.



Fig. 2. Variation of break thermal efficiency vs load.

2) Specific Fuel Consumption:

Figure 3 depicts the Brake specific fuel consumption of Diesel and Honge biodiesel blends at various loads. The characterization is that as load increases BSFC decreases and it is referred from the figure 3 that the BSFC of biodiesel blend (B20Honge+DEE+Ethanol) is more compared to that of Diesel at initial loads but it tends to be similar at higher loads due to the addition of oxygenated additives like DEE and Ethanol as they

are more volatile which boosts up the mixing velocity of fuel air mixture and results in good combustion process and they /exhibit marginal variation as load increases. The endorsement for the increase in BSFC of biodiesel blend (B20Honge+DEE+Ethanol) compared to Diesel is as Biodiesel has lower calorific value and higher viscosity which in turn produces less heat energy compared to Diesel and produces same energy output with more mass of fuel consumed.[7-9].



Fig. 3. Variation of brake specific fuel consumption vs load.

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B) Combustion Characteristics.

1) Heat release rate:

Figure 4 depicts the variation of heat release rate of the Biodiesel blends and Diesel at maximum loading conditions which nearly occurs near to the TDC (Top dead centre). The maximum heat release increases with addition of DEE and Ethanol concentration, which is seen at the end of compression at TDC [10]. DEE and Ethanol act as Oxygenator and improves the overall combustion for all fuel blends. The laminar flame propagation speed of Ethanol and DEE results in high heat release rates compared to that of Diesel. It can also be observed that the negative depression of the curve in the cycle which is during the ignition delay period and it occurs due to the loss of heat from the cylinder during the vaporization phase of the fuel and poor cetane value of Ethanol. The peak heat release rate observed from the graph is 86 J °/CA



Fig. 4. Variation of HRR versus crank angle(θ)

2) Cylindrical peak pressure:

Figure 5 depicts the variation of pressure in the combustion chamber at the various crank angles. It was found that the cylindrical peak pressure was increased at all load levels from 6 to 11 bars with blended fuel as compared to diesel. The validation for the increase in pressure is due to the improved

combustion of fuel and the presence of oxygen in the fuel particle enhances the low temperature reaction in proper direction[12].The maximum increase in pressure from 6 to 11 bars is for B20 Honge +Ethanol +DEE.



Fig. 5.Variation of pressure versus crank angle(θ).

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C) Emission Characteristics.

1) Carbon monoxide emissions:

Carbon monoxide (CO) is one of the fatal pollutant greenhouse gas and highly toxic.Figure 6depicts the emission of carbon monoxide of fuels tested with various loads. It is observed that the variation in CO emissions for all biodiesel blends and diesel is quite similar. It is also identified that Diesel emits more CO due to incomplete combustion of carbon. This may be due to the oxygen content and less C/H ratio of biodiesel that causes complete combustion. The biodiesel contains more oxygen which easily converts CO to CO₂. The CO emissions can be reduced by increasing oxygen percentage in charge. Initially the air fuel composition is a lean mixture hence the CO emissions are high but at half load the air fuel mixture ratios are optimum hence the CO readily converts into CO_2 but at high loads the air fuel ratios are of rich mixtures which results in improper combustion and leads to increase in co emissions.



Fig. 6.Variation of Carbon monoxide vs load.

2) NOx emissions:

The figure 7 depicts the variation of NO $_x$ emissions of Honge blends and Diesel over various loads. The NO $_x$ emissions for Biodiesel (Honge B20 + DEE + Ethanol) is higher compared to Diesel, as the lower heat release rate of Diesel due to lower combustion temperatures. As the Biodiesel blends (Honge B20 + DEE + Ethanol) contain high Oxygen the nitrogen does not readily react with oxygen at lower combustion temperatures, at

higher temperatures they react partially with Oxygen to produce Nitrogen oxides. This is due to the presence of antipenultimate shells of Nitrogen they tend to behave inert. NOx emissions for Diesel were relatively lower for all Start of ignition timings compared to Biodiesel blends. NOx emission of B20 biodiesel is found to decrease marginally as compared to Diesel. Due to higher HRR the in cylinder temperature would also increase thereby NOx emissions will be more[11].



Fig. 7 Variation of NOx emissions vs load.

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3) UBHC emissions:

Figure 8 exhibits the UBHC emissions with various loads applied. It is noticed that the UBHC variation is significantly varying for Biodiesel blend (B20Honge+DEE+Ethanol) with addition of Oxygenated additives. As load increases the emissions of the biodiesel blends at initial loads were quite more compared to Diesel due to less volatility and low temperatures. At higher loads the Biodiesel blends exhibited less emission compared to Diesel due to the presence of oxygen in biodiesel and higher combustion temperature, which promotes the oxidation of hydrocarbon emissions. The reason for this behaviour of diesel is because it is highly volatile compared to biodiesel which makes a large difference in the emissions between Diesel and Biodiesel at higher loads. Cetane number also plays an important role as Biodiesel blends have higher cetane number compared to Diesel which reduces ignition delay. Further the oxygen content of DEE and Ethanol leadsto low emission[11,12].



Fig 8. Variation of UBHC emission vs load

4. CONCLUSION

Based on the analysis of different blends of diesel with honge oil the following conclusions have been drawn:

- The oxygenated additives such as DEE and Ethanol improved atomization, combustion and showed better emission characteristics through higher engine brake power, lower BSFC and higher BTE.
- Diethyl ether showed the improvement through its density, viscosity properties and high calorific value but ethanol showed less development because of less calorific value and poor cetane number.
- Decrease in CO and UHC emissions due to the higher oxygen content and NO decreased for high latent heat of evaporation.
- At half load, emission of CO is reduced in the blend containing B20 Honge+DEE+Ethanol compare to pure diesel.
- We can conclude that oxygenated additives are very important for biodiesel and also experiments are still going on improving calorific value, cetane number and other properties.

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