

Exploring Inorganic Phosphate Solubilizing Trait of Halotolerant Rhizobacteria Isolated From *Cuminum cyminum*

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Abstract—Phosphorus is an important macronutrient that is essential for plant growth and development. Inorganic phosphorus (P), which can make up to 70% of the total P content in soils, can exist in calcium-, aluminum- or iron-complexed forms that are unavailable for plant use. As a result, mineral phosphorus, P₂O₅, is often used as a fertilizer to supplement the nutrient for crop growth. Soil microorganisms play a role in maintaining the ecological balance by active participation in nutrient cycles in nature. Phosphate Solubilizing Bacteria (PSB) as a phosphorus bio fertilizer improves soil fertility by solubilizing insoluble phosphate salts and increase crop production. This research aimed in isolation salt-tolerant or halotolerant PSB will facilitate the development of saline-alkali soil-based agriculture. The fastest growth was observed when the culturing temperature was 30°C±2°C and the concentration of NaCl was 6% (w/v). It was found that the isolates can survive at a concentration of NaCl up to 20%. Halo zone formation and plate screening method was used for phosphate solubilization evaluation on Pikovskaya's agar medium, containing insoluble tri-calcium phosphate (TCP). Isolates showing highest Phosphate Solubilization Index (PSI) were selected for further study as qualitative as well as quantitative activities. The growth of the isolates was related with a significant decrease of pH of the medium which indicates the production of acids like acetic acid, formic acids etc. These isolates will be further identified by 16S rRNA, their morphological and biochemical properties respectively.

Keywords: Phosphorus, PSB, NaCl, Phosphate.

1. INTRODUCTION

Phosphorus is an important plant nutrient next only to nitrogen. It is a master key element in crop production (Pierre, 1938). It is a component of key molecules such as nucleic acids, phospholipids and ATP. It stimulates the root growth at early stages of plant growth, helps in early maturity of crops and inadequate availability of it markedly reduces plant growth, reduces plant size and unusually deep green colouration. Plants acquire P from soil as phosphate anions. P is abundant in soils in both organic and inorganic forms, P deficiency is a serious concern for agriculture productivity and sustainability around the globe. Despite having rich total P contents, the available P in even most fertile soils are too low to meet most plants' demands due to precipitation with Ca²⁺ in alkaline soils (Rahmatullah et al. 1994) and with Al³⁺ and Fe³⁺ oxides in acid soils (Plaxton and Carswell 1999; Raghothama 1999)

Since phosphorus content is quite low in saline soil so the application of phosphatic fertilizer is essential for better crop yield. However, P reacts very strongly with soil constituents, as a result only about 5–25% of applied P can be utilized by the plants while the remaining 75–95% of applied P is being rapidly fixed and transformed into insoluble forms unavailable to plants (Yin et al. 2015). More than 2/3 of this fertilizer is rendered unavailable due to fixation in soil complex. They also cause surface and ground water pollution and various other deleterious effects.

Several reports indicated that various bacterial species have the ability to solubilize insoluble inorganic phosphate compounds like Tricalcium Phosphate (TCP), Dicalcium Phosphate (DCP), hydroxyapatite and Rock phosphate. Solubilization of insoluble P by microorganisms was reported by Pikovskaya (1948). These specific groups of soil microbes known as

PSB(Phosphate Solubilizing Bacteria) which increase the availability of phosphates to plants, not only by mineralizing organic phosphorus compounds but also by rendering inorganic phosphorus compounds more available to them.(Gaur,1979). PSB vary in efficiency in solubilizing inorganic phosphate are known to render insoluble phosphate into a soluble form through the process of solubilization (inorganic P)/mineralization (organic P) by releasing low molecular mass organic

acids, phosphatases or other complex agents (Duponnois et al. 2005). The concentration of iron ore, temperature, and C and N sources greatly influence the P-solubilizing potentials of these microbes. Among the various nutrients used by these microorganisms, ammonium salts has been found to be the best N source followed by asparagine, sodium nitrate, potassium nitrate, urea and calcium nitrate (Ahuja et al. 2007).The role of PSB is shown in **Figure:1**.

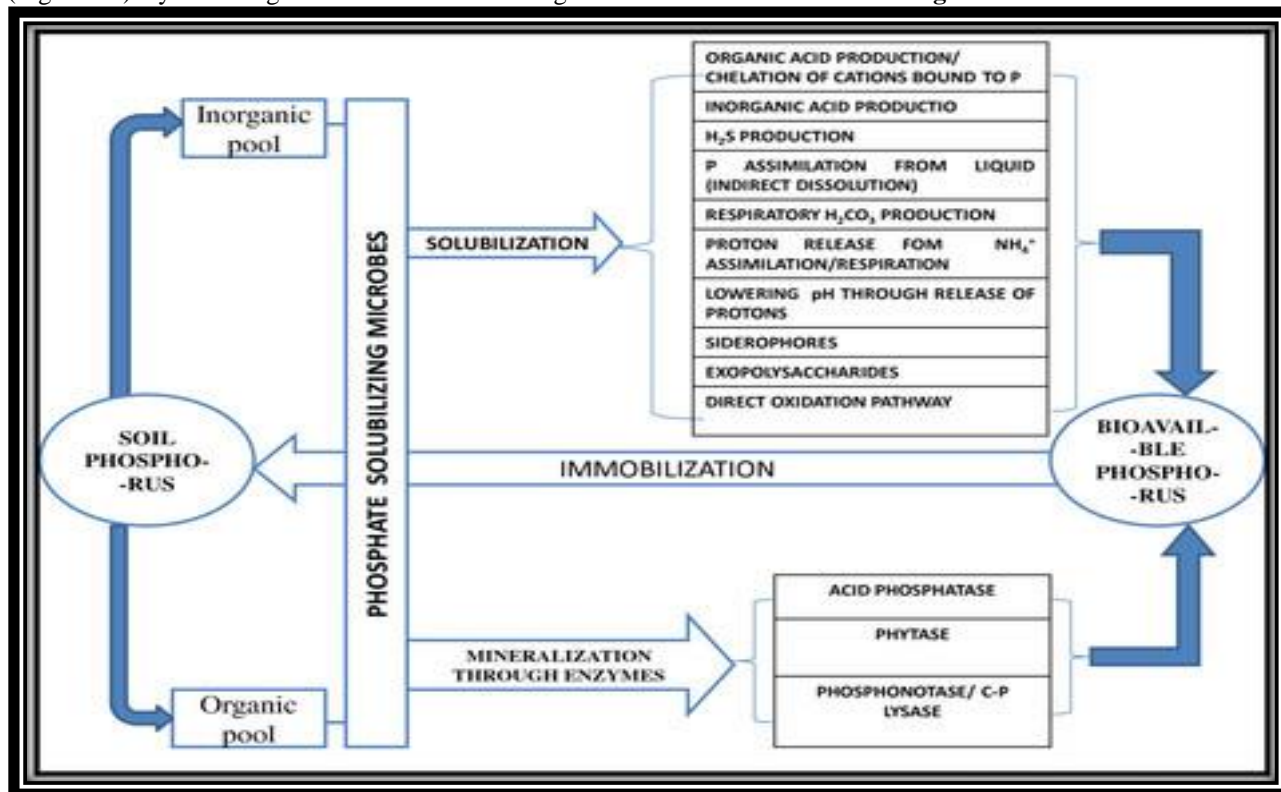


Figure: 1 Schematic representation of PSB mechanism in increasing soil phosphorus (Seema et al.2013)

Application of phosphate solubilizing microorganisms including bacteria, fungi and yeast to agriculture ecosystems have been considered as a low-cost and low-energy management strategy to enhance the agronomic effectiveness and efficiency of both soluble and insoluble P fertilizers, especially insoluble RP (Biswas and Narayanasamy, 2006; Xiao et al. 2013; Abbasi et al. 2015; Bakhshandeh et al. 2015). PSB constitute 1-50%. Moderately halotolerant bacteria are capable of growing at 4 to 20%(0.8 to 3.4 M) NaCl concentration. These halotolerant bacteria are capable to solubilize phosphate in saline areas and make it available to plants in saline areas. They are the promising strategy to improve world food production without causing any environmental hazards.

2. MATERIALS AND METHODS

2.1 Study sites and rhizospheric soil sampling

3 different sites from Kachchh district , Gujarat,India were selected for sampling of soil in 2015. The air temperature of the sites varied between 25°C and 30°C, soil temperature 20°C and 25°C. Before sampling, grass, forest litter or any other material on the soil surface were removed. Within each site, rhizospheric soil samples of *Cuminum cyminum* from 15 cm were collected by soil auger and mixed as one composite sample and stored at 4°C for isolation of PSB. A subsample of about 0.5 kg was air-dried and passed through 2-mm sieve and used for the determination of physical and chemical characteristics. The physicochemical parameter of the samples were

analysed at the sampling site (*in situ*) and in the microbiology laboratory (*ex situ*) using the physicochemical methods mentioned below.

The physical parameters includes:

1. pH by Ion Specific Electrode method
2. Electric Conductivity (Kalra and Maynard, 1991)
3. Salinity (Electrical salinity method)
4. Bulk density (Laboratory method for disturbed soil)
5. Soil texture (Laboratory method for disturbed soil)
6. Soil moisture by Gravimetric method
7. Redox potential by Potentiometric method

2.2 Isolation of halotolerant rhizobacteria

Composite Rhizospheric soil sample stored at 4°C was used for bacterial isolation. Rhizospheric bacteria were isolated by adding 1g soil sample in 99 ml sterile distilled water and kept the flask on shaking for 20 min to separate microorganisms completely from the soils, allowed to stand for 1 hour. Then the supernatant was used for serial dilution plating on nutrient agar plates having NaCl concentration of 6%. The plates were incubated at 30 ± 2°C till the appearance of bacterial colonies. Individual colonies were picked and streaked on nutrient agar plates for further purification. Morphological characteristics of colonies were recorded after 24 h of growth on nutrient agar plates at 30 ± 2°C. Bacterial colonies having different morphological characteristics were screened and 30 isolates were selected for evaluating their cell morphology. The slides of screened strains were prepared to study Gram positive or negative characteristics by staining techniques (Murray et al. 1994).

2.3 Culture Maintenance

Bacterial cultures were maintained on nutrient agar slants having 6% NaCl concentration as per in their respective plates and preserved at 4°C under refrigerator

2.4 Phosphorus Solubilization Assay

A) Qualitative Assay

Phosphorus solubilizing capacity of isolates on Pikovskaya's agar plate medium having bromophenol blue dye (Gupta et al, 1992) and NaCl concentration upto 6%. The potential of isolates to solubilize insoluble

phosphates was examined on the Pikovskaya's medium (Pikovskaya 1948). Each bacterial culture in its isolated single colonial form was spot inoculated in the center of agar plates of Pikovskaya's media, containing tricalcium phosphate [Ca₃(PO₄)₂] as insoluble phosphate source. The plates were incubated at 28 ± 2°C for 6 days. Isolates capable of solubilizing insoluble phosphates will form halos. Using the diameter of clearing halozones, P solubilization index (PSI) was calculated by the formula mentioned below (Edi-Premono et al. 1996).

$$\text{PSI} = \frac{\text{colony diameter} + \text{halozone diameter}}{\text{colony diameter}}$$

Enrichment Method for Isolation

1g of soil sample was added to 100ml Pikovskaya's broth containing flask and three successive transfers were made weekly intervals to enrich the medium. Then from the final flasks, PSBs were isolated by streaking a loopful of the culture on solid phosphate medium. The colonies showing halo zones were purified and preserved.

B) Quantitative Assay

Phosphorus solubilizing capacity of PSB was determined in 100 mL Pikovskaya's broth medium (liquid medium) containing TCP insoluble [Ca₃(PO₄)₂]. The insoluble P source was added to 100 mL broth before sterilization. Then 1 ml suspension of each bacterial culture was added to the broth under sterilized conditions. One control i.e. one without PSB or plant growth promoting rhizobacteria (PGPR) was also maintained. The PSB were allowed to grow for 21 days at 28 ± 2°C on rotatory shaker. Growth medium (5ml) was withdrawn aseptically at 7 days interval from each flask and centrifuged at 10,000rpm for 20 mins. The supernatant was analyzed for P₂O₅ content by Chlorostannous reduced molybdophosphoric acid blue method using Systronic 166 spectrophotometer (Gaur, 1990). The pH of the supernatant was also checked by pH meter (and correlated with the results obtained on plates and in liquid medium. Standard graph was prepared using KH₂PO₄ (2-10ppm/ml) as standard reagent. Figure:2 showing the protocol followed for isolation of phosphate solubilizer. The isolates will be further identified by 16S r-RNA gene analysis. Further phylogenetic analysis will be used to confirm that these isolates were closely related to their respective strains through MEGA 7.0 software.

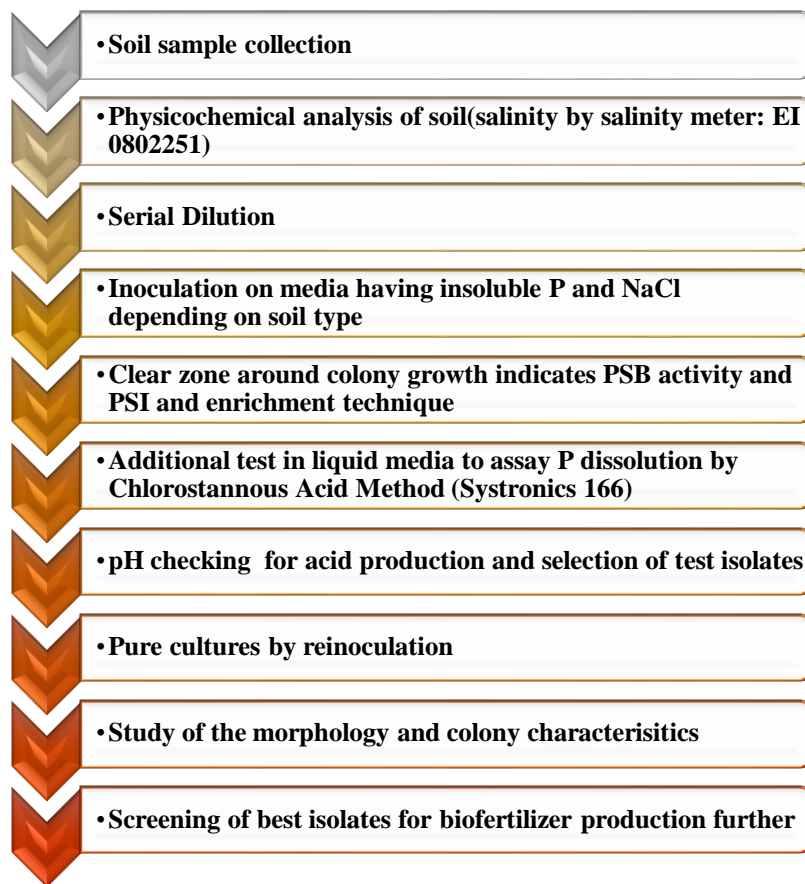


Figure:2 Protocol followed for isolation of phosphate solubilizer

3. RESULTS AND DISCUSSION

3.1 Physical Characteristics of soil sample

Physical characteristics of soil sample of cumin rhizosphere like colour, appearance, soil type, temperature (both internal and external) were studied. The soil colour is brown with salt crystals and its texture was sandy to loamy. The outer temperature of the soil was $40^{\circ}\text{C} \pm 2^{\circ}\text{C}$ whereas the internal temperature was $35^{\circ}\text{C} \pm 2^{\circ}\text{C}$.

3.2 Chemical Characteristics of soil sample

Soil sample showed the higher value of Cl^- , redox potential, salinity and Fe content with highest TDS

values (Table 1). Higher value of redox potential reflects higher microbial activities in the soil. Highest Tds and pH showed that the soil was alkaline in nature. There was presence of other soluble salts also that helps in making the soil saline in nature. This indicated that soil is not suitable for the farming. Higher nitrogen content was found in the soil sample indicated that the soil have higher microbial counts and diversity. Zhou et al. 2002 suggested that the organic and inorganic content of soil is one of the most important key for driving the microbial community structure.

Table:1 Physicochemical analysis of soil

Sr. No.	Parameter	Unit	Cumin
			Soil
1	pH	Direct	8.6
2	Electric Conductivity	Milli.moh/cm	0.60
3	Org – C	In %	1.37

4	Nitrogen	In %	0.12
5	Av. - P	ppm	3.62
6	Av. - K	ppm	26
7	Cu	ppm	1.82
8	Zn	ppm	2.32
9	Mn	ppm	5.94
10	Fe	ppm	58.75
11	Boron	ppm	1.34
12	Sulphur	ppm	21.7
13	Cl ⁻	mg/K	300
14	TDS		250
15	mV		252
16	Salinity	ppt	50
17	Bulk density	mg/m ³	1.33
18	Soil moisture	In%	12.60

3.2 Isolation of halotolerant rhizobacteria

Isolates showing variations in their colony morphology were selected and purified by subculturing. Total 30 isolates were screened finally for Phosphate solubilization under saline condition (Figure:3). Only

those isolates which gave maximum growth (fast growers), zone of phosphate solubilization were selected for further studies. Colony characteristics of isolates growing on 6% NaCl concentration is shown in Table:2

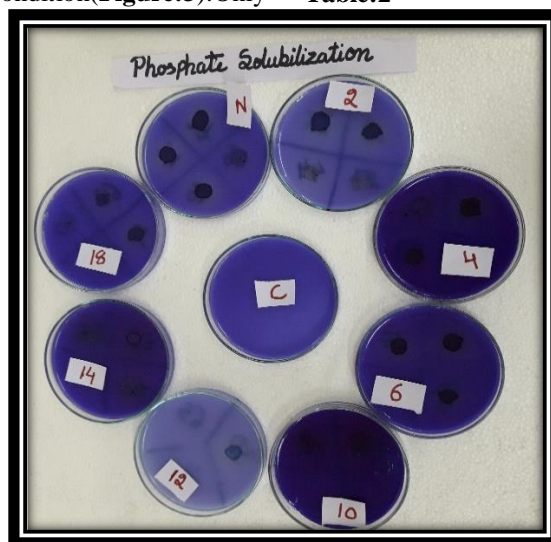


Figure:3 Isolates on Pikovskaya agar plates having NaCl concentration ranging from 2% to 18%

Table:2 Colony characteristics of isolates growing on 6% NaCl concentration

Characteristics	Type of colony on NA(6%NaCl)			
	3	8	18	31
Size	small	medium	small	medium
Shape	round	round	round	round
Margin	even	even	even	even
Surface	rough	smooth	rough	smooth

Elevation	flat	raised	flat	raised
Pigmentation	off white	pale white	off white	pale yellow
Consistency	dry	butyrous	dry	butyrous
Opacity	opaque	opaque	opaque	opaque
Gram's Reaction	positive	negative	positive	negative

3.3 Qualitative Assay

Results indicates that all the 4 four isolates showed zone of phosphate solubilization on solid Pikovskaya's medium having 6% NaCl concentration after 3 d of incubation at $30\pm 2^\circ$ (Figure :4) The PSI of all four isolates ranging from 6 to 12.

3.4 Quantitative Assay

Results indicates that all the 4 four isolates showed phosphate solubilization in liquid Pikovskaya's medium having 6% NaCl concentration after 14 d of incubation at $30\pm 2^\circ$ (Figure: 5.) that was measured spectrophotometrically by Chlorostannous acid method.

Change in pH of the medium from 6.5 to a maximum of 3 was observed (Table:3). This change indicates that there is production of some organic acids that would contributed in inorganic phosphate to solubilize and make it available in soluble form. The phosphate solubilization in liquid medium was found to ranging from $9.66 \mu\text{g/ml}$ to $20 \mu\text{g/ml}$ under saline condition of 6% NaCl concentration.

These all indicates that all the four isolates are capable of growing under saline conditions as well as can solubilize phosphate efficiently and can make available to plants for their better growth and production.

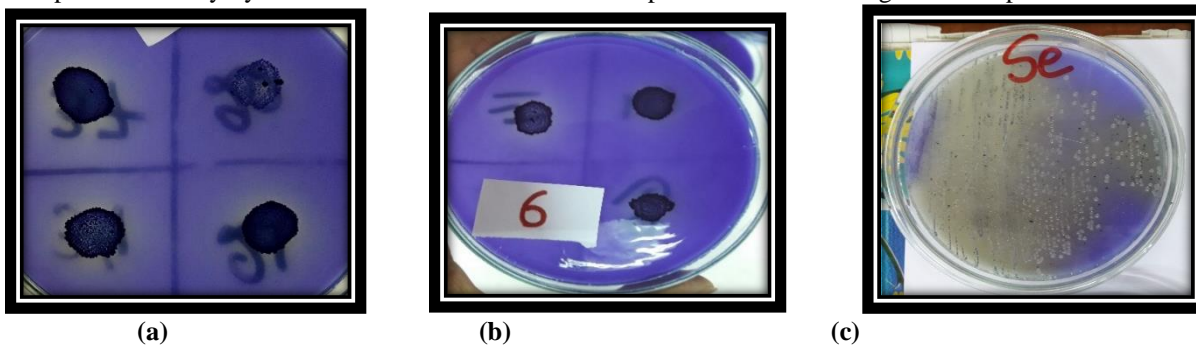


Figure :4(a) and (b) yellow zone of phosphate solubilization on Pikovskaya's agar plates having 6% NaCl (c) Enrichment technique plate



Figure: 5 Quantitative estimation of phosphorus solubilized by PSB

Table: 3 Change in pH of the medium and PSI of isolates

Isolate No.	pH 7 th d	pH 14 th d	PSI
3	5.5	3.7	6

8	5.2	3.0	10
18	5.4	3.3	8
31	5.0	3.3	12

4. CONCLUSION

From this study it is concluded that isolates which gave maximum growth (fast growers) ,zone of phosphate solubilization growing on 6% NaCl concentration indicates their tolerance towards salt stress. They can be used as bio fertilizers in saline areas for crop production. Halotolerant PSB will facilitate the development of saline-alkali soil-based agriculture .The PSI of all the studied isolates was in range of 6 to 12 which implicates that these PSB can be used as a means of fixing the nutrient availability in the soil especially phosphorus.This will reduce the use of chemical PNK fertilizers for higher crop yield. Phosphorus is an important macronutrient that is essential for plant growth and development. They will involve in maintaining the phosphorus cycle. They can easy shows symbiotic association with plants roots.PSB function is in long duration, causing improvement in soil fertility.Using these isolates as bioinoculant will make salt affected area to support the cultivation of crops . They can maintain the natural habitat of soil. They are easy to apply without any additive agents. They are cost effective bio fertilizer and having low manufacturing cost. Hence,PSB can be considered as an ecofriendly approach towards sustainable agriculture path.

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